Abstract No. Chid0377

Grazing Angle Infrared Spectroscopy of Chromate Conversion Coatings

D. Chidambaram, G. Halada and C. Clayton (SUNY, Stony Brook) Beamline(s): U10B

Introduction: Dichromate based solutions have been used as corrosion inhibitors on aluminum since the beginning of the last century (1,2). Since then, the mechanism of protection offered by chromate-based solutions has been under investigation. Toxicity and mutagenic effects of hexavalent chromium have stimulated the development of alternative coatings (3-4), the success of which depends on understanding precisely the underlying mechanism of protection offered by hexavalent chromium. The ability of Fourier transform infrared spectroscopy to be operated in various modes makes it a very attractive analytical technique.

Methods and Materials: Polished AA2024-T3 (1 μ m finish, Cu-free polishing solution) samples were conversion coated using Alodine 1200S® (Parker Amchem) for 300 seconds. The samples were rinsed thoroughly in deionized water followed by drying with nitrogen. Infrared spectra were collected at UV beamline U10B using a Thermo-Nicolet FT-IR Continuum microscope in both grazing angle (GAIRS) and near normal reflectionabsorption (RAIRS) modes.

Results: Figs. 1 & 2 show the spectra obtained from RAIRS and GAIRS modes respectively. The strong and broad absorption band centered around 3600-3000 cm $^{-1}$ is due to both OH stretching and water of crystallization (5). The sharp spectral feature observed at ~2080 cm $^{-1}$ is due to the C-N stretching mode from the Fe(CN) $_6$ 3 -. Chromate conversion coatings contain a significant amount of cyanide. HOH bending of lattice water is observed at ~1620 cm $^{-1}$. A very weak absorption due to organic amines may also be seen centered around 1410 cm $^{-1}$. Absorption due to hexavalent chromate and dichromate occurring at ~920-950 cm $^{-1}$ and ~820-840 cm $^{-1}$ corresponding to Cr(VI)-O vibration are also labeled. All the above spectral features are routinely observed on CCCs and are discussed elsewhere (6-8). A spectral feature is also observed ~605-607 cm $^{-1}$ especially using the GAIRS method of analysis. We believe this band to be due to Cr₂O₃ as reported by Guillamet et al. (9).

It is observed from the spectra that the GAIRS method is more sensitive to the surface coating than the RAIRS method of analysis. This is clearly seen in the case of spectral features due to Cr(VI)-O vibration (~920 cm⁻¹), HOH bending of lattice water (1620 cm⁻¹) and the feature at ~606 cm⁻¹. Also the signal to noise ratio is higher in case of the GAIRS as can be seen from the high quality spectrum near the lower wavenumbers. In fact the spectral feature at 606 cm⁻¹ is almost masked in the reflection-absorption mode by the noise around that feature. Table 1 provides a comparison of peak heights as observed in each spectrum. The peak heights are then normalized with respect to the broad 3300 cm⁻¹ feature due to OH stretching. The peak heights have been used for comparison over peak area, as peak areas are dependent on peak width and may lead to erroneous results in case of peak width broadening as observed in case of the feature due to OH stretching. Table 1 indicates that the sensitivity can be increased by nearly an order of magnitude by the use of GAIRS as compared to the standard reflection-absorption method of analysis.

Conclusions: The sensitivity to chromate using a grazing angle objective was observed to increase by nearly an order of magnitude when compared to near-normal reflection-absorption infrared spectroscopy mode. Hence synchrotron radiation based grazing angle infrared spectroscopy can prove to be a highly effective tool in studying the local speciation or chemistry of surfaces under ambient conditions.

Acknowledgments: The U.S. Air Force Office of Scientific Research under contract F4962093-1-0218 supported this work. Lt. Col. Paul Trulove, Ph.D. served as contract officer. NSLS is supported by the U.S. Department of Energy under Contract No. DE-AC02-98CH10886. We thank Lisa Miller and Nebojsa Marinkovic for discussions.

References:

- 1. L. H. Callendar, Engineering, 120, 340 (1925).
- 2. W. Herbig, Aluminium, 19, 371 (1937).
- 3. C. M. Shepherd and R. L. Jones, U.S. Clearinghouse Fed. Sci. Tech. Inform., AD, 21pp (1971).
- 4. V. Bencko, Journal of Hygiene, Epidemiology, Microbiology and Immunology, Czechoslovakia, 29, 37 (1985).
- 5. G. Socrates, Infrared Characteristics Group Frequencies, John Wiley & Sons, New York, (1994).17.
- F. W. Lytle, R. B. Greegor, G. L. Bibbins, K. Y. Blohowiak, R. E. Smith, and G. D. Tuss, <u>Corrosion Science</u>, 37, 349 (1995).
- 7. L. Xia and R. L. Mccreery, Journal of the Electrochemical Society, 145, 3083 (1998).
- 8. D. Chidambaram, M. Jaime Vasquez, G. P. Halada, and C. R. Clayton, <u>Surface and Interface Analysis</u>, Vol. in press (2002).
- 9. R. Guillamet, M. Lenglet, and F. Adam, Solid State Communications, 81, 633 (1992).

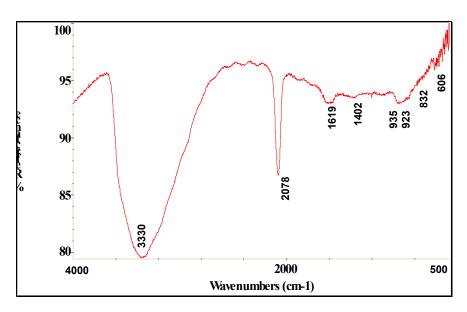


Fig 1. FT-IR spectrum of a chromate conversion coating obtained using reflection-absorption mode.

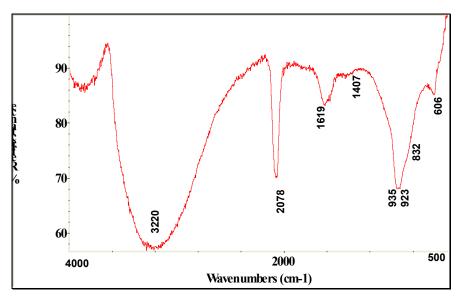


Fig 2. FT-IR spectrum of a chromate conversion coating obtained using grazing-angle mode.

Band	Normal Reflectance	Normalized w.r.t. H₂O	Grazing Angle	Normalized w.r.t. H₂O	Ratio of Grazing Angle /Normal Reflectance
H ₂ O	16.16	1.000	35.73	1.000	1.00
CN.	8.68	0.537	20.04	0.561	1.04
OH.	0.91	0.056	5.673	0.158	2.82
CrO ₄ ²⁻	0.95	0.059	19.44	0.544	9.22
606	0.29	0.018	2.898	0.081	4.50

Table 1: Peak heights, normalized peak heights and ratio of peak heights obtained by RAIRS and GAIRS techniques.